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To cite this article: Ramya Suseenthar, Palanivelan Ramachandran, Muthusamy Ranganathan, Mathiyazhagan Narayanasamy, Ramkumar Govindaraju, Prakash Balu & Nivetha Muthuganesh (23 Aug 2025): Biosynthesis, characterization, and insecticidal toxicity of iron oxide (Fe₂O₃) nanoparticles synthesized from marine macroalgae (*Ulva lactuca*), Toxin Reviews, DOI: [10.1080/15569543.2025.2548215](https://doi.org/10.1080/15569543.2025.2548215)

To link to this article: <https://doi.org/10.1080/15569543.2025.2548215>



Published online: 23 Aug 2025.



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RESEARCH ARTICLE



Biosynthesis, characterization, and insecticidal toxicity of iron oxide (Fe_2O_3) nanoparticles synthesized from marine macroalgae (*Ulva lactuca*)

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ABSTRACT

Background and aim: The use of synthetic pesticides presents environmental and health hazards, necessitating the development of environmentally sustainable alternatives. This study seeks to synthesize and assess the insecticidal effectiveness of iron oxide nanoparticles obtained from the green seaweed *Ulva lactuca* against different instars of lepidopteran and dipteran insects.

Experimental approach: The synthesis of Fe_2O_3 nanoparticles was conducted via *U. lactuca* extract, followed by characterization via FTIR, XRD, TEM, SEM, and EDS studies. Larvicidal bioassays were performed at concentrations between 50 and 200 $\mu\text{g/mL}$ for 48 hours targeting second to fourth instar larvae of *Spodoptera frugiperda*, *Spodoptera litura*, *Helicoverpa armigera*, *Tuta absoluta*, *Aedes aegypti*, and *Culex quinquefasciatus*.

Key findings: FTIR analysis showed Fe–O bonding at 668cm^{-1} , while XRD revealed distinct diffraction peaks at 2θ values of 43.2° , 50.1° , and 62.1° , thereby substantiating the crystalline structure of Fe_2O_3 . TEM and SEM demonstrated consistently spherical nanoparticles ($\sim 100\text{nm}$), while EDS verified iron as the predominant element. The Fe_2O_3 nanoparticles demonstrated significant dose-dependent larvicidal efficacy, with maximum mortality rates of 96.22% in *H. armigera* ($\text{LC}_{50} = 216.76\mu\text{g/mL}$), 96.10% in *T. absoluta* ($\text{LC}_{50} = 116.36\mu\text{g/mL}$), and 97.0% in *Ae. aegypti* ($\text{LC}_{50} = 136.81\mu\text{g/mL}$).

Conclusions: The findings indicate the significant and extensive insecticidal efficacy of *U. lactuca*-synthesized Fe_2O_3 nanoparticles, endorsing their application as an environmentally benign nanobiopesticide.

IMPACT STATEMENT

This research illustrates the effective green manufacturing of Fe_2O_3 NPs utilizing *U. lactuca*, providing a sustainable substitute for chemical insecticides. The synthesized nanoparticles shown significant larvicidal efficacy against major agricultural and public health insect pests, such as *Spodoptera*, *Helicoverpa*, *Tuta*, *Aedes*, and *Culex* species. This sustainable method combines marine resources with nanotechnology to create efficient, biodegradable, and nontoxic biopesticides. The results offer considerable potential for integrated pest management approaches, decreasing reliance on synthetic pesticides and alleviating its environmental and health repercussions. This research promotes the development of marine algal nanobiotechnology in sustainable agriculture and vector management.

ARTICLE HISTORY

Received 17 March 2024

Revised 19 July 2025

Accepted 8 August 2025

KEYWORDS

Biofabrication; *Ulva lactuca*; insecticides; vector mosquitoes; biocontrol agents

1. Introduction

Insect pests present considerable risks to agriculture and public health (Belluco *et al.* 2023). Mosquitoes are notably perilous vectors, disseminating several life-threatening diseases that impose significant health and economic burdens globally, particularly in underdeveloped countries (Mohan *et al.* 2024). The two principal mosquito genera, *Aedes* and *Culex*, facilitate the

transfer of pathogens responsible for diseases including dengue, malaria, chikungunya, lymphatic filariasis, and yellow fever, impacting both humans and animals (Dahmana and Mediannikov 2020, Hillary *et al.* 2021). *Aedes aegypti* is a principal vector for the dengue virus, with more than 5.2 million documented cases in 2019. Mortality rates have significantly escalated, particularly among younger demographics from 960 deaths in 2000 to 4,032 in 2015 (WHO 2022). Lymphatic filariasis

impacts almost 80 million individuals each year, with *Culex quinquefasciatus* responsible for nearly 30 million of these instances. In India, around 19 million individuals exhibit symptoms of filarial illness, adding to a worldwide total of 120 million cases (Ramkumar *et al.* 2023).

Agricultural pests significantly affect crop output and quality in numerous places. Significant insect pests, including *Spodoptera frugiperda*, *Spodoptera litura*, *Helicoverpa armigera*, and *Tuta absoluta*, inflict considerable harm on over 180 plant species, encompassing economically vital crops such as cotton, tomato, cabbage, and cauliflower (Tarusikirwa *et al.* 2020, Khan *et al.* 2022, Lateef *et al.* 2024, Narayanan 2025). Historically, pest management has predominantly depended on synthetic chemical insecticides (Kandil *et al.* 2020). Nonetheless, overutilization has resulted in numerous issues, such as the emergence of insecticide resistance, environmental pollution, and adverse impacts on non-target organisms (Arokiyaraj *et al.* 2014, Ali *et al.* 2022, Siddiqui *et al.* 2022, Khangura *et al.* 2023, Muthusamy *et al.* 2024). These disadvantages highlight the pressing necessity for environmentally sustainable and efficient pest management options.

Nanotechnology has emerged as a new application in the biosciences, which shown effective against insect pests by infiltrating plant tissues and serving as carriers for active substances (Khot *et al.* 2012, Nie *et al.* 2023, Narayanan and Prabhu 2025, Narayanan 2025, Kadry *et al.* 2025). Studies indicate that silver and graphene oxide nanoparticles can interfere with insect detoxification enzymes, leading to oxidative stress and cellular death (Altammar *et al.* 2023, Basheer *et al.* 2025). Metal-based nanoparticles, including Fe_2O_3 , disturb insect physiology by attaching to sulfur and phosphorus in proteins and nucleic acids, thereby compromising membrane integrity.

The green synthesis of nanoparticles has attracted interest as a sustainable alternative to chemical synthesis. It improves the efficacy and stability of active chemicals in pesticides while reducing environmental risks (Gonzalez-Merino *et al.* 2021, Gupta *et al.* 2023). Fe_2O_3 nanoparticles are notably appealing owing to their biocompatibility, magnetic characteristics, ease of retrieval, and efficacy in biological systems (Mittal *et al.* 2022, Nadeem *et al.* 2022). In contrast to chemical approaches, green synthesis employs natural plant or algae extracts to decrease and stabilize nanoparticles, offering an environmentally sustainable and scalable approach for nanoparticle manufacturing (Kiwumulo *et al.* 2022, Bensy *et al.* 2022, Liu *et al.* 2023, Osman *et al.* 2024, Mani *et al.* 2024). These extracts are abundant in biomolecules, including flavonoids, phenols, and

polysaccharides, which facilitate nanoparticle synthesis (Radulescu *et al.* 2023). Marine algae, including *Ulva lactuca*, are being increasingly investigated for these purposes (Tonelli *et al.* 2023, Kandasamy *et al.* 2023). This species possesses a plethora of bioactive substances, including as sulfated polysaccharides, proteins, and phenolics, which promote the production and stabilization of nanoparticles (Alboofetileh *et al.* 2021, Mohan *et al.* 2024). *Ulva lactuca* demonstrates significant antibacterial and insecticidal characteristics, positioning it as a viable choice for the eco-friendly production of Fe_2O_3 nanoparticles intended for pest management (Rima *et al.* 2021, Vijayakumar *et al.* 2025, Salem *et al.* 2019). Despite its extensive availability and biological potential, limited research has investigated its utilization in the synthesis of Fe_2O_3 nanoparticles for insecticidal purposes, namely targeting agricultural pests and mosquito vectors.

This research fills the gap by synthesizing Fe_2O_3 nanoparticles with *U. lactuca* extract and assessing their insecticidal effectiveness on several pest species and developmental stages. The nanoparticles were analyzed by UV-Vis spectroscopy, Fourier Transform Infrared Spectroscopy (FTIR), X-ray diffraction (XRD), scanning electron microscopy (SEM), and transmission electron microscopy (TEM) to ascertain their size, shape, and functional groups. Insecticidal experiments were performed on relevant agricultural pests and mosquito vectors to evaluate toxicity and efficacy. The research seeks to position green-synthesized Fe_2O_3 nanoparticles as a feasible, environmentally acceptable substitute for traditional chemical pesticides in sustainable pest control.

2. Materials and methods

2.1. Green algae

The *U. lactuca* was chosen for its well-documented characteristics and year-round availability. It was collected from Gulf of Mannar coastal region at the depth of 2–6 meters. The chosen seaweed was identified by Dr. V. Veeragurunathan principal scientist, CSIR-CSMSRI Marine Algae Research Station, Mandapam Campus, Tamil Nadu, India.

2.2. Aqueous extract preparation

The collected seaweed samples were immersed in a container of tap water and subsequently cleaned twice with distilled water to eliminate unwanted contaminants (Figure 1). The purified algae were shade-dried for 15 days at ambient temperature ($25 \pm 1^\circ\text{C}$) with relative humidity of 50–60% prior to being pulverized



Figure 1. Seaweed *Ulva lactuca* used for the iron oxide nanoparticles synthesis.

into a fine powder using a mechanical blender (L5M-A, Silverson, USA). Approximately 25 g of seaweed powder was mixed with 250 mL of distilled water and heated to 90°C for 30 min. The extract was filtered using Whatman no. 1 filter paper and preserved at refrigeration temperature of 4°C till subsequent usage.

2.3. Green synthesis of Fe_2O_3 -NPs

The iron oxide nanoparticles were synthesized according to the procedure outlined by Mahdavi *et al.* (2013) with some changes. Approximately 50 mL of *U. lactuca* extract was combined with an equal volume (50 mL) of 0.4 M FeCl_3 solution. The mixture was thoroughly agitated and positioned on the hot plate (100°C) with continuous magnetic stirring for 45 min to monitor color changes. The sample was centrifuged at 5,000 rpm for 20 min at 4°C to eliminate any residual aqueous extract from the freshly synthesized NPs. The aqueous mixture containing Fe_2O_3 -NPs was centrifuged a second time and resuspended in double distilled water. Finally, the Fe_2O_3 -NPs were ultimately dissolved in 100% ethanol prior to characterization.

2.4. Characterization of Fe_2O_3 -nanoparticles

The biosynthesized seaweed iron oxide nanoparticles were characterized using several analytical techniques, including UV-visible spectroscopy, Fourier transform infrared spectroscopy (FT-IR), X-ray diffraction analysis (XRD), scanning electron microscopy (SEM), and transmission electron microscopy (TEM) and energy-dispersive X-ray analysis (EDXA). The preliminary verification of Fe_2O_3 -NPs synthesis was performed utilizing UV-visible spectrometry to determine the spectral frequency range of 350–800 nm and the peaks were identified. The biomolecules responsible for the reduction of ferric chloride were determined by FT-IR

analysis carried out in the range of 400–4,000 cm^{-1} (Thermo Nicolet Nexus 670). The primary elemental composition of the Fe_2O_3 -NPs was ascertained using energy-dispersive X-ray (EDX) using X-ray micro-analyzer (shimadzu XRD, 6000). Hitachi H-600 scanning electron microscope (SEM and TEM) was used to examine the surface morphology, size and shape of the synthesized Fe_2O_3 -NPs. EDXA was employed to ascertain the elemental composition of the sample at the acceleration voltage of 20.0 kV.

2.5. Insect culture and rearing

The egg masses of *S. litura* and *H. armigera* (Accession no.NBAII-MP-NOC-01) were obtained from National Bureau of Agricultural Insect Resources live insect repository (NBAIR), Bangalore, Karnataka, India. The egg masses of *S. frugiperda* and *T. absoluta* were collected from tomato and maize fields. The samples were confirmed with Dr. Asokan R, Principal Scientist, ICAR and maintained in IIHR (Indian Institute of Horticulture Research, Bangalore, Karnataka, India). Newly molted larvae were reared on artificial diet consisting 159 g chickpea flour 150 g, 0.75 g sorbic acid, 24 g yeast powder, flax oil 6 mL, agar 8.4 g, vitamin mixture 0.02 g (thiamine, pyridoxine, and biotin), ascorbic acid 2.35 g, methyl-4-hydroxy benzoate 1.5 g, distilled water 550 mL, and 0.75 g streptomycin, in accordance with the established protocol (Ahmad *et al.* 2018). *A. aegypti* and *Cx. quinquefasciatus* egg masses were provided by the Institute of Vector Control and Zoonoses (IVCZ, Hosur, Tamil Nadu, India). The newly emerged mosquito larvae were reared in a plastic tray (30×20 cm) covered with mosquito netting, with dog biscuits provided as their nutrition. All insect cultures were sustained at $27 \pm 1^\circ\text{C}$ temperature with relative humidity of $70 \pm 5\%$ under 10:14 h light: dark photoperiod condition in PG and Research Center in Biotechnology laboratory until the appearance of adulthood.

Adult male and females were released in disposable transparent plastic jars (15 cm × 15 cm × 30 cm) at 1:3 ratios for mating and were provided with a 10% sugar solution for sustenance. Female mosquitoes were provided chick blood in 60×60 cm rearing cages and fresh trays were installed as ovipositor. Following hatching, uniformly sized 2nd, 3rd, and 4th instars larvae were chosen for the larval bioassays.

2.6. Artificial diet bioassay

The artificial diet bioassay was performed using the established protocol of Jafir *et al.* (2021) with few alterations. Small diet samples (5 mm³) were manually prepared and immersed in varying concentrations of *Ulva*

Fe₂O₃ NPs (50, 100, 150, and 200 µg/mL) before being air-dried for 20 min. Simultaneously, early 2nd, 3rd, and 4th instar larvae of each insect underwent a 1-h fasting before being introduced to their introduction into the treatment diet. Untreated larvae were maintained as a control group, receiving only water. Twenty-four hours following treatment, small quantities of untreated meals were administered to satisfy the requirements of the insect larvae. The experiment was conducted under optimal conditions, as previously mentioned, with three replicates for each concentration to minimize error. The larval mortality was assessed 48 h after treatment.

2.7. Mosquito larvicidal bioassay

The larvicidal effectiveness of particular mosquito species was assessed with slight alterations (Aziz 2023). Newly hatched early 2nd, 3rd, and 4th instar larvae were placed into a 250-mL paper cup with varying concentrations of *Ulva* Fe₂O₃ prepared for treatment (50, 100, 150, and 200 µg/mL). Larvae introduced into distilled water served as a control. Each dosage was reproduced thrice, with 25 similarly sized larvae introduced into each paper cup. After 24 h, the larvae were supplied with dog biscuits for nourishment. The larval mortality was assessed 48 h after treatment.

2.8. Data analysis

The data collected from the aforementioned experiments underwent statistical analysis, with values shown as their mean ± SE. The toxicity data were evaluated using factorial analysis of variance (one-way ANOVA) and Dunnett's multiple comparison test $P < 0.05$ for pairwise mean comparisons. Probit analysis was utilized to ascertain the lethal concentrations at LC₅₀ and LC₉₀ values employing the SPSS statistical software, Version 16.0 (SPSS Inc., Chicago, IL, USA).

3. Results

3.1. Synthesis of Fe₂O₃-NPs

This study concentrated on devising a biosynthetic approach for generating iron oxide nanoparticles utilizing an aqueous extract of *U. lactuca*. Figure 2(A, B) illustrates that the color of the FeCl₃/*Ulva* extract solutions at ambient temperature swiftly transitioned from yellow to dark brown, signifying the synthesis of Fe₂O₃ nanoparticles within the seaweed extract. The incorporation of FeCl₃ into the *Ulva* extract may yield sulfated polysaccharides as a predominant constituent, which,

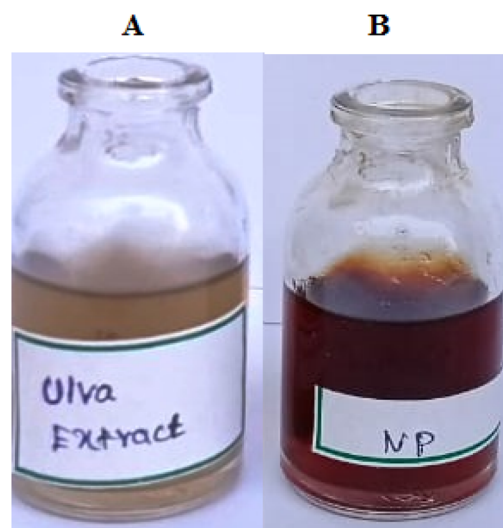


Figure 2. Fe₂O₃ NPs synthesized from *Ulva lactuca* aqueous extract. (A) Crude extracts. (B) Iron oxide nanoparticles (Brown color).

due to the presence of sulfate, hydroxyl, and aldehyde groups, could facilitate the reduction of Fe³⁺ and the stabilization of nanoparticles.

3.2. Characterization of Fe₂O₃-NPs

3.2.1. Ultraviolet-visible spectroscopy (UV-vis)

The UV-Visible spectra of aqueous extract of *U. lactuca* and synthesized Fe₂O₃-NPs are shown in Figure 3. The SPR band 1 indicates near 402 nm is a synthesized iron oxide NPs, confirming their role in the color shift and the production of Fe₂O₃-NPs. Peak 2 near 450 nm may indicate the presence of impurities in the sample.

3.2.2. Infrared spectroscopy

FTIR was employed to ascertain the functional groups of the active components by analyzing the peak values in the infrared radiation spectrum. Figure 4(A, B) illustrates the FTIR spectra of *Ulva* crude extract and Fe₂O₃-NPs synthesized using seaweed extract. The FTIR spectrum of *Ulva* extract exhibits distinctive stretching vibration bands at 668 cm⁻¹ and 778 cm⁻¹, indicating the Fe-O stretch and validating the synthesis of Fe₂O₃. Supplementary peaks within 2850–2950 cm⁻¹ indicated C-H bonds, but a broad peak at 3400 cm⁻¹ revealed the existence of hydroxyl (OH) groups derived from *U. lactuca* bioactive chemicals that stabilize the NPs.

3.2.3. X-ray diffraction (XRD)

The crystalline structure and phase identification of nanoparticles were characterized by XRD. The XRD

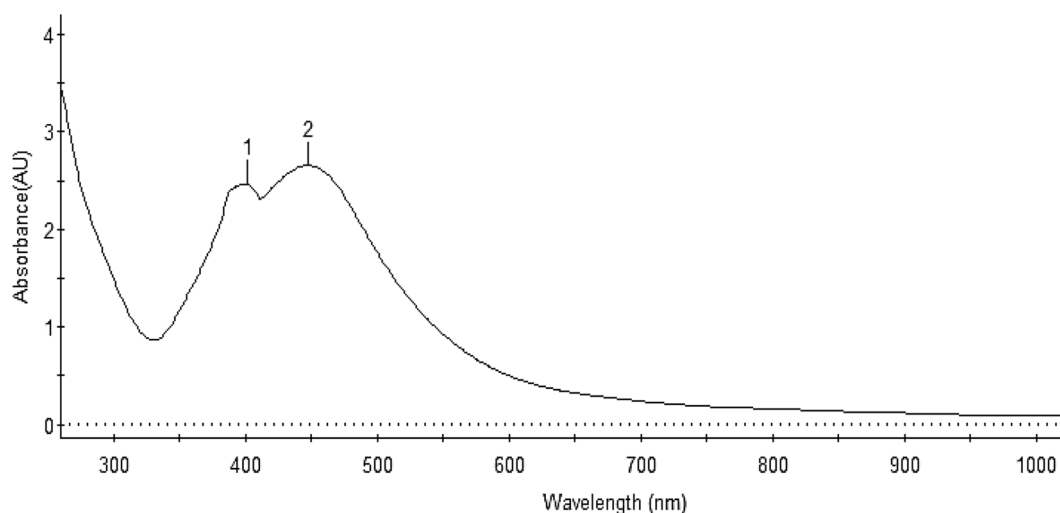
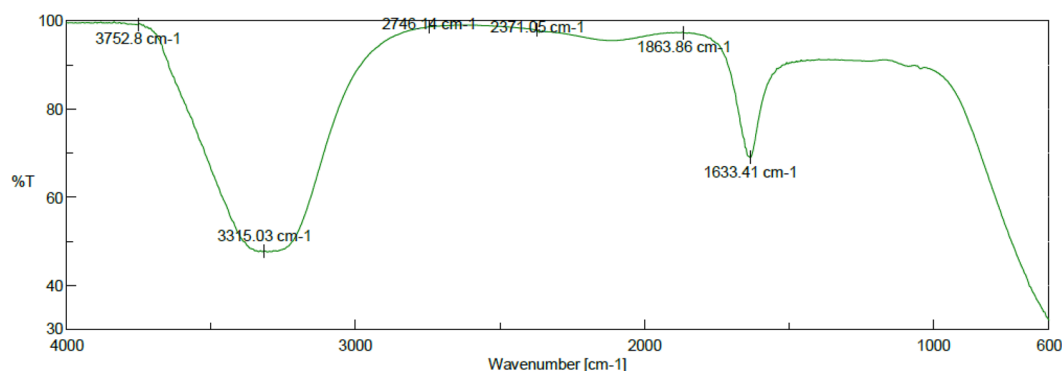
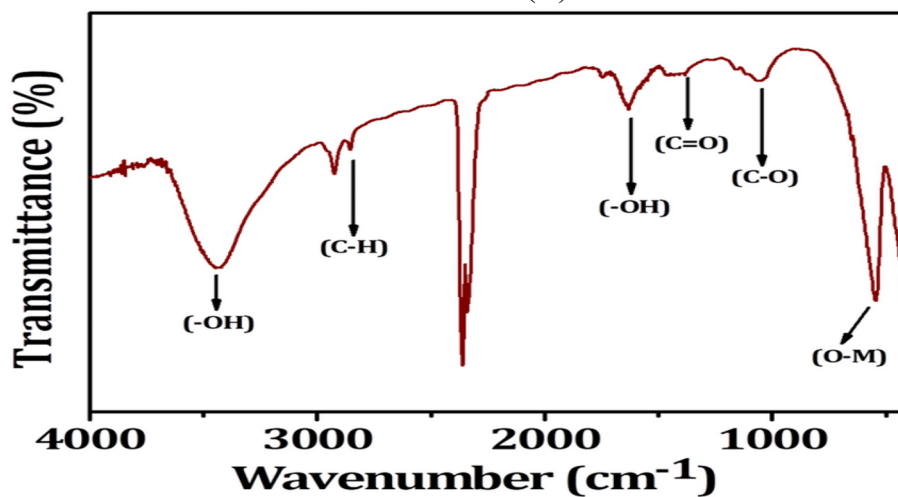


Figure 3. UV-Vis spectrophotometry showing the spectra of *U. lactuca* crude synthesized Fe_2O_3 NPs.



(A)



(B)

Figure 4. (A) FT-IR analysis of aqueous extracts of *Ulva lactuca*. (B) FT-IR with the peaks of *Ulva lactuca* and Fe_2O_3 nanoparticles.

pattern of the Fe_2O_3 NPs obtained from *U. lactuca* extract is shown in Figure 5. It is found that there exist strong diffraction peak with 2θ values of 25.1° , 43.2° ,

50.1° , 62.1° , 68.7° , 73.1° , 80.0° , and 82.1° , corresponding to the crystal planes of (104), (110), (192), (265), (300), and (341) of crystalline Fe_3O_4 -NPs, respectively.

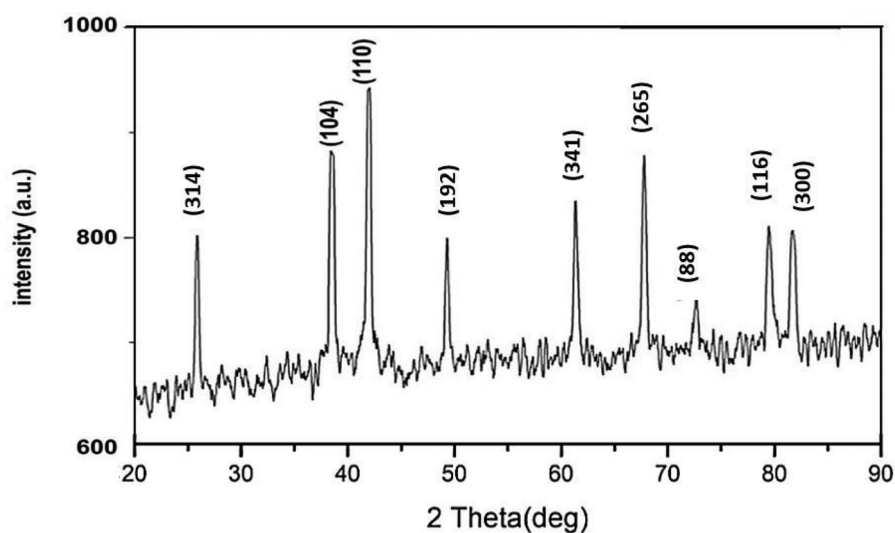


Figure 5. XRD pattern of Fe_2O_3 nanoparticles synthesized from *Ulva lactuca*.

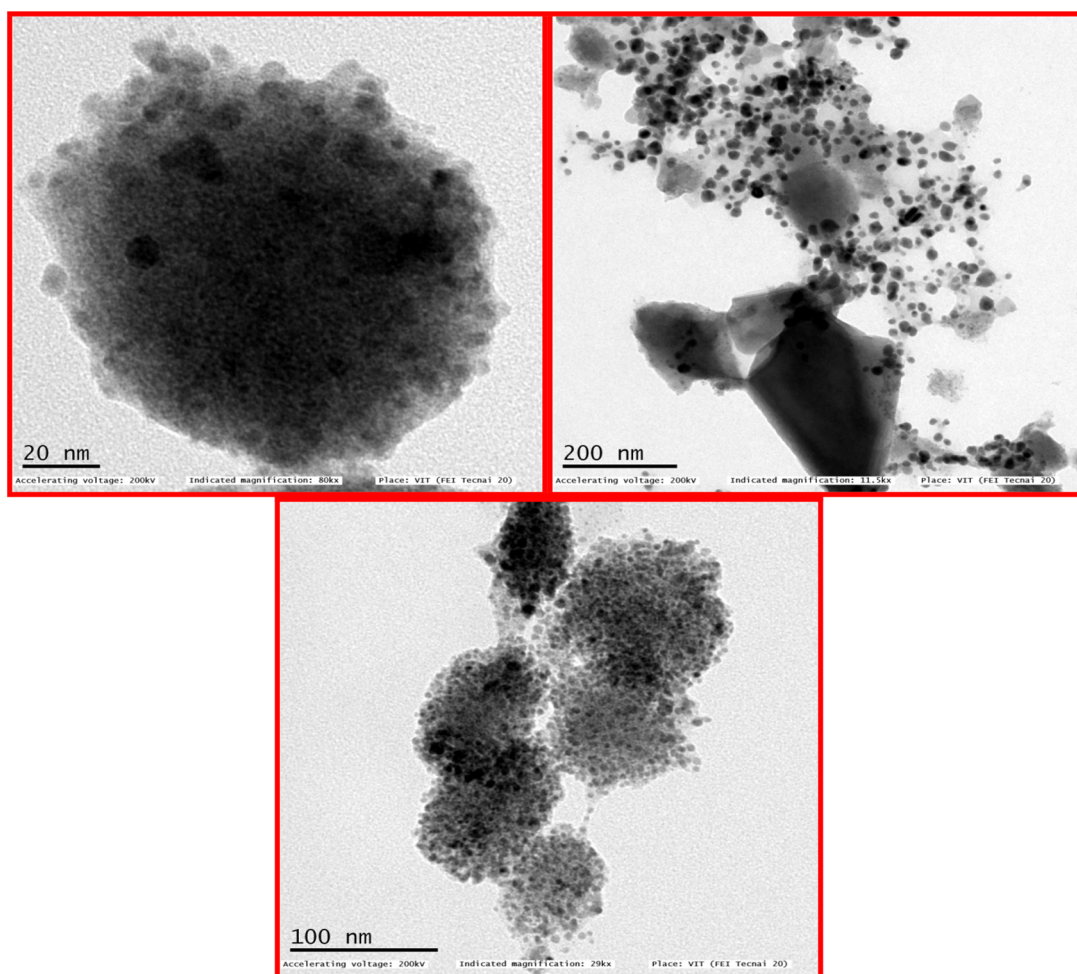


Figure 6. Transmission electron microscope (TEM) images of iron oxide nanoparticles from *Ulva lactuca* at different magnification range.

3.2.4. Transmission electron microscope (TEM)

For morphology analysis by TEM, a drop of Fe_2O_3 -NPs synthesized by treating FeCl_3 solution with *Ulva* extract was deposited on to a TEM copper grid. The images were

captured from the grid followed by drying. The images of TEM with different magnification size are shown in Figure 6. It is clear that from 100nm the sizes of the Fe_2O_3 -NPs are almost uniform and all of the particles are cubic in shape.

3.2.5. Scanning electron microscope (SEM)

The SEM image and EDaX spectra are shown in Figures 7 and 8. The SEM images confirm that the *Ulva*-derived Fe_2O_3 -NPs are cubic in shape. In the EDaX spectra, the

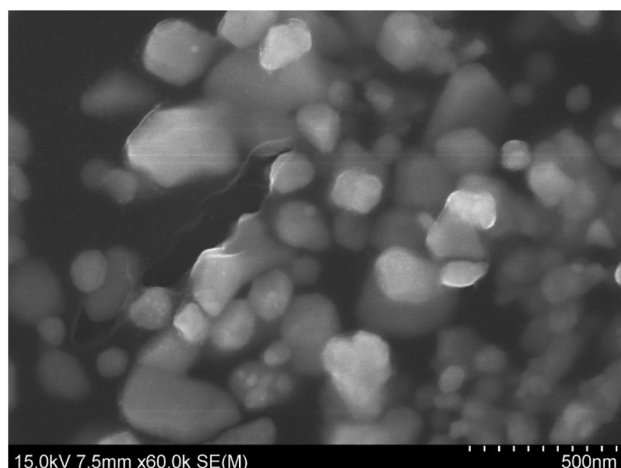


Figure 7. Scanning electron microscope (SEM) image of iron oxide nanoparticles from *Ulva lactuca*.

peaks around 0.8, 6.2, and 7.0 keV are related to the binding energies of Fe. Hence, the EDaX spectra of the Fe_2O_3 /*Ulva* extract confirmed the presence of Fe_2O_3 -NPs in the *U. lactuca* aqueous extract.

3.3. Mortality effect of *Ulva lactuca* Fe_2O_3 -NPs on lepidopteran insects

The toxicological impact of *U. lactuca* Fe_2O_3 -NPs against 2nd, 3rd, and 4th instar larvae of *S. frugiperda*, *S. litura*, *H. armigera*, and *T. absoluta* has been evaluated by artificial diet bioassays (Table 1.) Results showed that all the administered doses are showed increased mortality (Figures 9(A)–(C) and 10 A). The maximum concentration (200 $\mu\text{g}/\text{mL}$) of *U. lactuca* Fe_2O_3 -NPs over 2nd instar showed the highest larval mortality against *S. frugiperda*, *S. litura*, *H. armigera*, and *T. absoluta* (95.17%, 92.21%, 96.22%, and 96.10%) followed by 3rd (89.11%, 87.0%, 90.10%, and 90.20%) and 4th instars (88.76%, 84.21, 85.0%, and 85.42%).

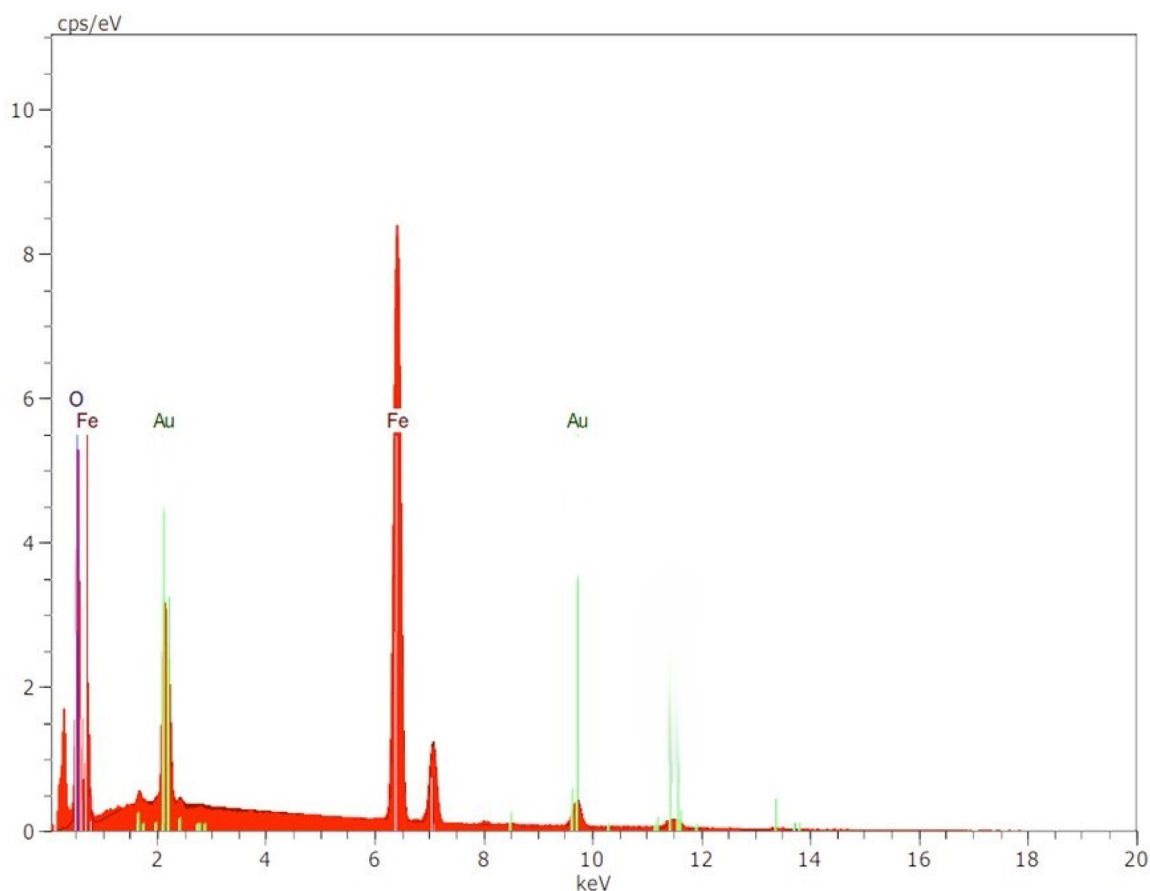


Figure 8. According to the EDaX image, the elementary composition evaluation ranges from 0 to 20 keV. Several elements are shown, and among the elements Fe (Iron oxide) is a major particle.

Table 1. Probit analysis for lethal concentration of iron oxide (Fe_2O_3) nanoparticles against different insects.

Insects	Larval Instar	$\text{LC}_{50} \pm \text{SE}$ ($\mu\text{g/ml}$)	$\text{LC}_{90} \pm \text{SE}$ ($\mu\text{g/ml}$)	Slope	χ^2
<i>S. frugiperda</i>	2nd	346.36 ± 33	1557.35 ± 2.44	0.462 ± 0.03	2.33
	3rd	425.40 ± 45	2174.54 ± 0.32	0.493 ± 0.07	7.15
	4th	532.43 ± 21	2721.21 ± 0.24	0.423 ± 0.07	7.20
<i>S. litura</i>	2nd	412.24 ± 44	2575.25 ± 5.41	1.212 ± 0.23	6.33
	3rd	577.31 ± 54	3144.36 ± 0.14	1.238 ± 0.30	0.40
	4th	622.43 ± 21	3421.11 ± 0.33	0.218 ± 0.67	1.62
<i>H. armigera</i>	2nd	216.76 ± 14	397.25 ± 24.29	1.921 ± 0.23	0.44
	3rd	397.09 ± 57	1126.55 ± 2.32	0.362 ± 2.07	0.91
	4th	412.22 ± 11	2127.36 ± 1.43	4.221 ± 0.37	5.20
<i>T. absoluta</i>	2nd	116.36 ± 41	197.10 ± 25.56	2.911 ± 0.23	1.08
	3rd	321.19 ± 67	1216.23 ± 7.72	2.200 ± 0.26	0.86
	4th	301.02 ± 0.9	1527.36 ± 4.13	1.112 ± 0.33	2.20
<i>A. aegypti</i>	2nd	136.81 ± 57	198.56 ± 16.44	2.606 ± 0.40	0.95
	3rd	295.47 ± 7.5	412.53 ± 25.65	1.993 ± 0.23	0.24
	4th	302.21 ± 5.5	521.21 ± 23.24	0.423 ± 0.07	3.04
<i>Cx. quinquefasciatus</i>	2nd	143.61 ± 18	323.10 ± 32.54	1.492 ± 0.30	4.39
	3rd	259.43 ± 22	474.54 ± 21.32	0.442 ± 0.27	6.15
	4th	273.81 ± 6.5	371.73 ± 20.24	0.332 ± 0.27	6.24

LC_{50} and LC_{90} lethal concentration that kills 50% and 90% of the test animal; SE, standard error; $\mu\text{g/ml}$, microgram per milliliter; χ^2 , chi-square.

3.4. Mortality effect of *Ulva lactuca* Fe_2O_3 -NPs on diptera insects

The mortality of *U. lactuca* synthesized Fe_2O_3 nanoparticles on *A. aegypti* and *C. quinquefasciatus* larvae is seen in Figure 10(B, C). The larval mortality was affected by increasing concentrations (50–200 $\mu\text{g/ml}$). High mortality was reported at elevated quantities of synthesized nanoparticles, with the rates of 97.0% and 93.80%. Statistical analysis of the bioassay results indicated a significant difference in larval mortality ($P < 0.05$).

4. Discussion

This study synthesized Fe_2O_3 -NPs utilizing *U. lactuca* as the major source. The synthesized NPs were characterized using various analytical techniques, including UV-spectrophotometer, FTIR, XRD, TEM, SEM, and EDAX. Bio-synthesized nanopesticides effectively enhance plant pest and vector management by accurately delivering active components to specific targets in a controlled manner (Zahin et al. 2020, Vijayakumar et al. 2025). This sophisticated method provides effective solutions and reduces pesticide usage (Eldesouky et al. 2023). The observed color transition from yellow to dark brown in this work signifies the reduction of ferric chloride to iron oxide nanoparticles, culminating in the synthesis of Fe_2O_3 -NPs. Fe_2O_3 nanoparticles are synthesized when the phytocomponents of seaweed extract act as a reducing agent on ferric chloride, therefore reducing ferric iron and yielding very stable iron nanoparticles (Bouafia and Laouini 2021, Mohanasundari et al. 2023).

The UV-Vis spectroscopy examination of crude and *U. lactuca* synthesized Fe_2O_3 -NPs are observed within

the spectral range of 402 and 450 nm. UV-Vis denotes the absorption spectrum within the UV-Vis spectrum, utilizing light in the near UV and near infrared ranges. In support of our findings, Mahdavi et al. (2013) found that the aqueous extract of *Sargassum muticum* facilitated the synthesis of Fe_3O_4 -NPs at UV wavelengths of 402 and 415 nm. Salem et al. (2019) identified the absorbance peaks ranging from 402 to 415 nm for the aqueous extracts of brown seaweed (*Colpomenia sinuosa*) and red sea weed (*Pterocladia capillacea*), whereas the Fe_2O_3 -NPs synthesized from *Apis mellifera* honey were measured at 350 nm (Shahid et al. 2023). The variations in SPR biogenesis of Fe_2O_3 -NPs can be attributed to the aggregation properties of *U. lactuca* nanoparticles of varying sizes and shapes under the given conditions (Azizi et al. 2013).

The FTIR examination of *U. lactuca* synthesized iron oxide nanoparticles revealed unique peaks (Figure 4(A, B)), indicating the Fe-O stretch and validating the creation of Fe_2O_3 , as well as the stabilization of nanoparticles by the bioactive compounds contained in *U. lactuca* (Azizi et al. 2013). This corroborates the mechanism in which phenolic and polysaccharide groups function as both reducing and capping agents (Sahayaraj et al. 2020). Comparable types of bands were seen in Fe_2O_3 nanoparticles biosynthesized from the aqueous extracts of the seaweed *C. sinuosa* (Salem et al. 2019).

The crystalline structure of Fe_2O_3 -NPs was examined using an X-ray diffractometer. Numerous diffraction peaks were seen in the bio-fabricated iron oxide nanoparticles (Figure 5). The synthesized nanoparticles exhibited distinct diffraction peaks at (314), (104), (110), (192), (341), (265), (116), and (300) indicating their well-defined crystalline structure. The XRD

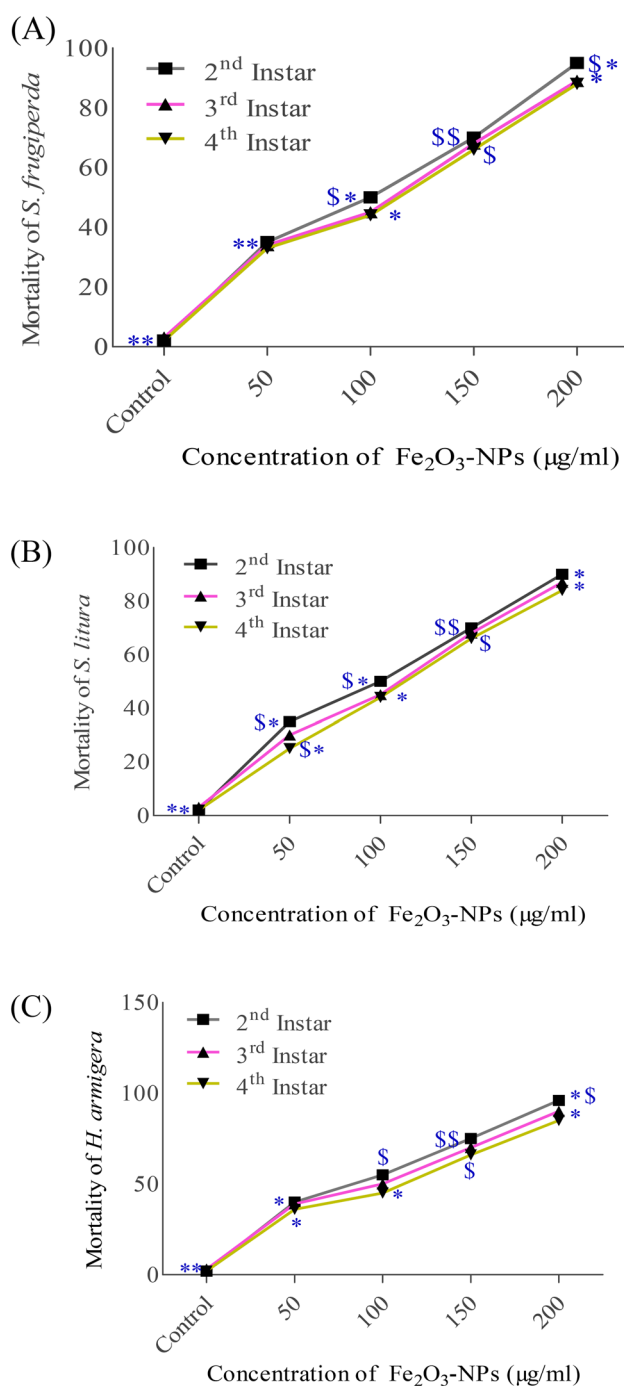


Figure 9. Percent mean mortality of *Ulva lactuca* Fe₂O₃ against *S. frugiperda* (A), *S. litura* (B), *H. armigera*, (C) 48-h post-treatment. Symbols with similar appearance are statistically significant at $P < 0.05$, different symbols are not significant (Dunnett's multiple comparison tests).

pattern obtained in this study aligns with previous findings on the XRD patterns of Fe₂O₃-NPs reported by Nadeem *et al.* (2022) and Ahmadi and Izanloo (2023). A significant diffraction peak was seen between 40° and 68° (2θ), confirming the ability of *U. lactuca* to synthesize iron oxide nanoparticles. Comparable outcomes in structural parameters, such as dislocation

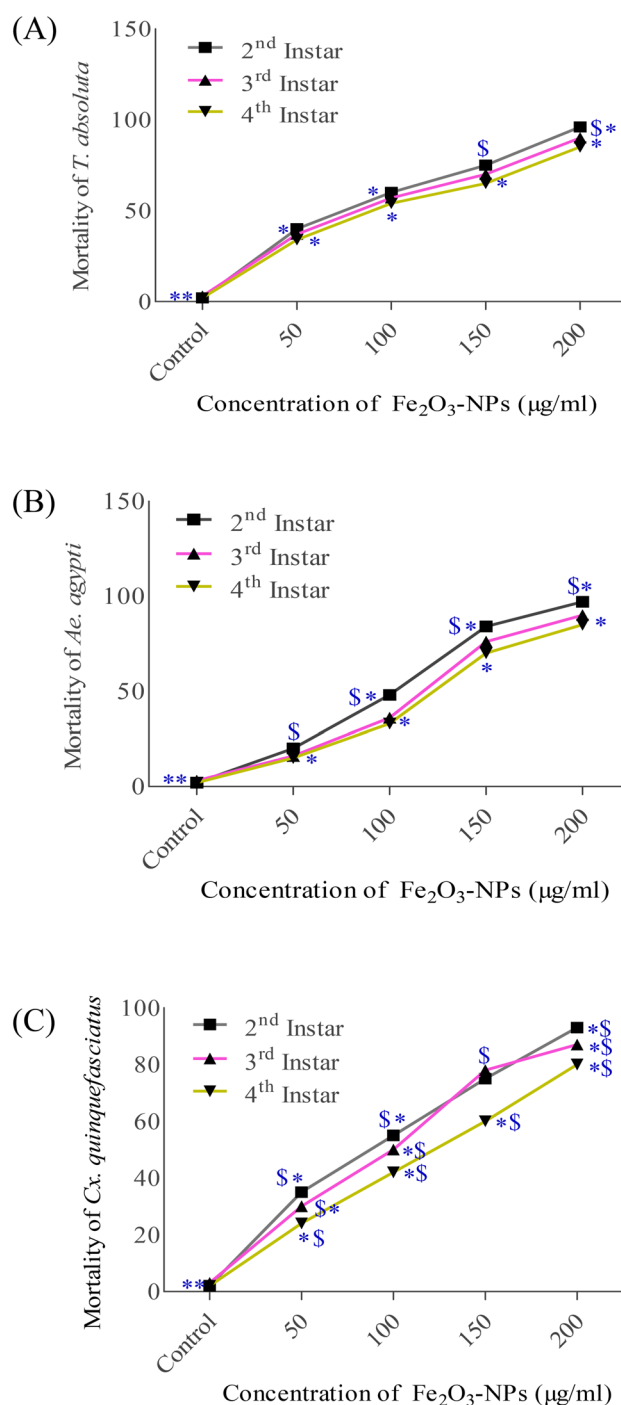


Figure 10. Percent mean mortality of *Ulva lactuca* Fe₂O₃ against *T. absoluta* (A), *Ae. aegypti* (B), *Cx. quinquefasciatus*, (C) 48-h post-treatment. Symbols with similar appearance are statistically significant at $P < 0.05$, different symbols are not significant (Dunnett's multiple comparison tests).

density and lattice constants, are consistent with the previous studies on biosynthesized Fe₂O₃-NPs (Chan *et al.* 2022).

TEM and SEM were utilized to ascertain the form and dimensions of the synthesized nanoparticles derived from *Ulva* seaweed. The current investigation established that the iron oxide nanoparticles produced

were consistent in size and exhibited spherical and cubic shapes. Prior research indicates that the majority of nanoparticles produced using green synthesis exhibit a spherical morphology (Murugan et al. 2015, Ahmad et al. 2020). Bhuiyan et al. (2020) revealed the heterogeneous particle size distribution of iron oxide nanoparticles synthesized from *Carica* papaya leaf extract. This heterogeneity may arise from the accumulation of minute constituents of diverse bioactive reducing agents derived from plant extracts.

The toxicological impact of *U. lactuca* Fe₂O₃-NPs on 2nd, 3rd, and 4th instar larvae of agricultural pests and mosquitoes was evaluated. The peak concentration (200 µg/mL) of *U. lactuca* Fe₂O₃-NPs during the 2nd instar exhibited the highest larval mortality rates against *S. frugiperda*, *S. litura*, *H. armigera*, and *T. absoluta*. The increased mortality rate observed in this study may be ascribed to the interaction of Fe₂O₃-NPs involved in metabolic activity and protein synthesis in insects (Kalpana et al. 2024). A similar study demonstrated that the aqueous extract of *Nephrolepis exaltata* produced iron oxide nanoparticles exhibiting notable anti-plasmodial activity (Nadeem et al. 2022). Jafir et al. (2021) validated our results by showing that a higher concentration of *Ocimum basilicum* silver nanoparticles resulted in increased mortality rates throughout different instars of *S. litura*.

Comparable to agricultural pests, green synthesized Fe₂O₃-NPs utilizing *Grevillea robusta* leaf extract exhibited significant mortality rates against 2nd and 4th instar *Ae. aegypti* (64–96%) and *An. stephensi* (56–84%) mosquito larvae, with LC₅₀ values of 238.05 ppm and 292.72 ppm, respectively (Zargham et al. 2023). Our research revealed an LC₅₀ of 136.81 and 143.61 µg/mL for *Aedes* and *Culex* larvae, respectively. Prior research demonstrated that silver nanoparticles synthesized from the macroalgae *C. spongiosus* caused a substantial mortality rate in *An. stephensi*, *Ae. aegypti*, and *Cx. quinquefasciatus*. The iron oxide nanoparticles synthesized from *Grevillea robusta* leaf extract shown mortality rates of 64–96% for *Ae. aegypti* (2nd and 4th instars) and 56–84% for *Anopheles stephensi*, with LC₅₀ values of 238.05 ppm and 292.72 ppm, respectively (Kumar et al. 2021, Li et al. 2022, Lippi et al. 2023, Zargham et al. 2023). Our work documented reduced LC₅₀ values of 136.81 µg/mL for *Ae. aegypti* and 143.61 µg/mL for *Cx. quinquefasciatus*, indicating the enhanced efficacy of *U. lactuca*-mediated Fe₂O₃-NPs.

5. Conclusion

This study efficiently demonstrated the synthesis of Fe₂O₃-NPs using the green seaweed *U. lactuca*. The

synthesized Fe₂O₃ nanoparticles exhibited significant dose-dependent larvicidal effectiveness ($P < 0.05$) against four agricultural pests and two mosquito vectors at various developmental stages. Among the assessed stars, the second instar larvae demonstrated the highest mortality rate, coupled with the lowest LC₅₀ values, followed by the third and fourth instars. The findings highlight the potential of *U. lactuca*-derived Fe₂O₃ nanoparticles as an eco-friendly and cost-effective biocontrol agent for lepidopteran and dipteran insect species.

Acknowledgments

We are grateful to PG and Research Centre in Biotechnology and Department of Microbiology, M.G.R College, Hosur, Tamil Nadu, India, for providing a research platform. We wish to acknowledge Periyar University, Salem, Tamil Nadu, India, for technical support.

Author contributions

The research was conceptualized by all of the authors. The original draught of the manuscript was overseen by S.R and R.M. SR was in charge of keeping daily track of the project's development. R.M, M.N, AS, HS, NP, and GR assisted in the execution of the experiment as well as the interpretation of the findings. All authors read and approved the manuscript.

Availability of data and material

On reasonable request, the corresponding author will provide the datasets created and/or analyzed during the current work.

Disclosure statement

The authors declare that they have no competing interests.

Funding

The author(s) reported there is no funding associated with the work featured in this article.

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