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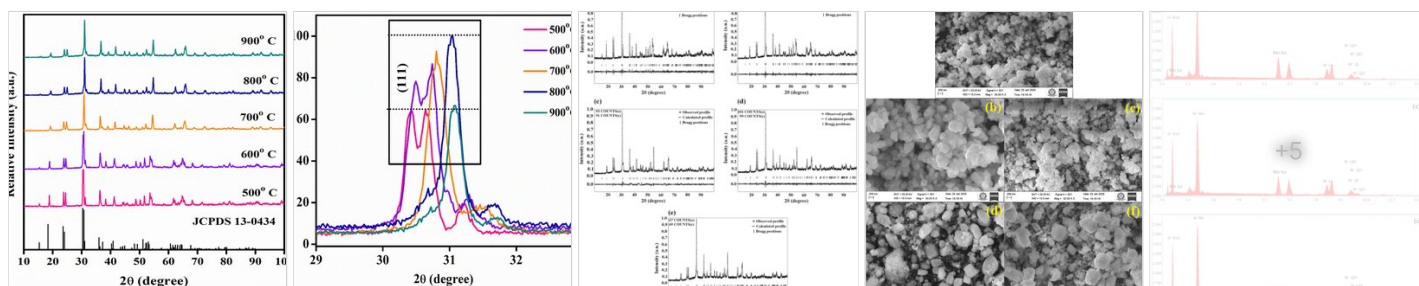
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Abstract and figures

The growing concern of drug pollution in water bodies, particularly the presence of pharmaceutical drugs like Diclofenac (DF), has prompted the emergence of photocatalytic degradation as a promising solution, driving the need for efficient photocatalysts to mitigate potential risks to aquatic ecosystems and human health. In this study, the influence of temperature on the degradation of DF (name of the drug) using MnWO₄ (manganese tungstate) as a photocatalyst is investigated. The precise co-precipitation method was used to synthesize MnWO₄, which was subsequently calcined at different temperatures ranging from 500 °C to 900 °C. The physicochemical properties of synthesized materials were investigated by various analytical and spectroscopical techniques. Significantly, MnWO₄ calcinated at 800 °C demonstrated exceptional photocatalytic performance, achieving a degradation rate exceeding 98% for DF under visible-light illumination. This superior activity can be attributed to factors such as excellent crystallinity, a well-defined morphology, a superior optical band gap for effective utilization of visible light, and reduced particle size compared to other MnWO₄ materials. This work paves valuable insights into the temperature-dependent synthesis and properties of MnWO₄ as a photocatalyst for DF degradation. The exceptional photocatalytic performance observed at 800 °C highlights the potential of MnWO₄ as an efficient and environmentally friendly material for drug decomposition under visible-light conditions.



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for drug degradation

Lalitha Kamarasu^{1,2} · Satya Sree Nannapaneni² · Saravanavadivu Arunachalam³ · Padmapriya Arumugam⁴ · Naresh Kumar Katari⁵ · D. Sivaganesh⁶

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Abstract

The growing concern of drug pollution in water bodies, particularly the presence of pharmaceutical drugs like Diclofenac (DF), has prompted the emergence of photocatalytic degradation as a promising solution, driving the need for efficient photocatalysts to mitigate potential risks to aquatic ecosystems and human health. In this study, the influence of temperature on the degradation of DF (name of the drug) using MnWO₄ (manganese tungstate) as a photocatalyst is investigated. The precise co-precipitation method was used to synthesize MnWO₄, which was subsequently calcined at different temperatures ranging from 500 °C to 900 °C. The physicochemical properties of synthesized materials were investigated by various analytical and spectroscopical techniques. Significantly, MnWO₄ calcinated at 800 °C demonstrated exceptional photocatalytic performance, achieving a degradation rate exceeding 98% for DF under visible-light illumination. This superior activity can be attributed to factors such as excellent crystallinity, a well-defined morphology, a superior optical band gap for effective utilization of visible light, and reduced particle size compared to other MnWO₄ materials. This work paves valuable insights into the temperature-dependent synthesis and properties of MnWO₄ as a photocatalyst for DF degradation. The exceptional photocatalytic performance observed at 800 °C highlights the potential of MnWO₄ as an efficient and environmentally friendly material for drug decomposition under visible-light conditions.

Keywords MnWO₄ · PXRD · Rietveld method · Diclofenac · Degradation


1 Introduction

Energy shortages and environmental pollution have been widespread concerns over the past few decades [1, 2]. Nonetheless, the quick blossoming of the pharmaceutical

industry creates hazardous effluent during development and production and contaminates soil and water systems when discharged [3–5]. Among the pharmaceutical drugs, DF is a widely used medicine that reduces swelling (inflammation) and pain. It's used to treat aches and pains, as well as problems with joints, muscles, and bones such as rheumatoid arthritis, sprains, back pain, toothache, migraine, gout, and ankylosing spondylitis [6, 7]. DF drugs, known for their incomplete metabolism in the environment and adverse effects on humans and animals such as chronic and pulmonary toxicity, present a significant concern for researchers. Therefore, the search for unique and efficient materials with high photodegradation efficacy is a key focus. Photocatalytic materials capable of degrading persistent aromatic compounds, which do not degrade naturally in the environment, have gained considerable interest. Photocatalytic strategies offer an environmentally friendly solution by utilizing sunlight to convert organic materials into green compounds [8, 9]. This approach addresses the challenge of persistent pollutants and their potential hazards. By harnessing the power of photocatalysis, researchers aim to provide a sustainable approach for remediation and purification of contaminated

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We have synthesized different concentrations of graphitic carbon nitride (GCN) (20 %, 40 %, 60 %, and 80 %) decorated manganese oxide (Mn₂O₃) for photocatalytic degradation of the organic contaminant methylene blue (MB) in the current study. Powder X-ray diffraction (PXRD) research was used to investigate the crystalline nature of synthesized GCN-coated Mn₂O₃. Scanning electron microscopy (SEM) analysis was...

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Hydrothermal synthesis of CuMoO₄ photocatalyst for the photocatalytic degradation of methylene blue (MB) in the aqueous phase was performed under visible light illumination. The produced nanoparticle was characterized using X-ray Powder Diffraction (XRD), Fourier-transform infrared spectroscopy (FT-IR), Scanning Electron Microscopy (SEM), Ultraviolet-visible diffuse reflectance spectra (DRS-UV),...

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